

Enhanced remanence behavior in mechanically alloyed $SmCo₅$

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Abstract

Sm–Co powders produced by a mechanical alloying process were investigated in terms of their structure, morphology and magnetic properties. Remanence enhancement, with M //M, ratios up to 0.69 were observed and the high coercivity was found to be related to the development of the $SmCo_{\epsilon}$ phase. \circledcirc 1998 Elsevier Science S.A. All rights reserved.

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mixture of hard and soft phases have been the subject of material has not received the same attention as either the considerable interest. These materials, whilst being micro- NdFeB or SmFeN materials. structurally isotropic, exhibit an enhanced remanence as a In this paper we report on a study of mechanically result of magnetic exchange coupling between the hard and alloyed materials with compositions close to SmCo_{5} . soft magnetic phases. The materials can be produced either by melt spinning or by mechanical alloying, combined with a suitable heat treatment. **2. Experimental**

Mechanical alloying was first developed as a means of manufacturing oxide dispersion strengthened super alloys Mechanically alloyed samples were prepared from [1], but has more recently found applications in the samarium $(\sim 130 \text{ }\mu\text{m})$ and cobalt $(\sim 70 \text{ }\mu\text{m})$ powders of formation of amorphous alloys [2] and magnetic materials 99.9% purity. Starting compositions in the range Sm_xCo_5 ,
[3]. Although initially, mechanical alloying, like melt where $x=0.95-1.25$, were mechanically alloyed fo spinning, was used primarily as a method for fabricating min in an argon atmosphere of 1 bar using four 22 mm conventional isotropic magnetic materials which could steel balls in an ASI Uni Ball Mill. Powders were either be plastic bonded or used as a precursor for die subsequently heat treated in an argon atmosphere using a upset forging, its potential for producing exchange coupled conventional vacuum furnace. Structures were determined materials was not realised until later. The first evidence for by X-ray diffraction using a Philips PW1710 diffractomeexchange coupling was in NdFeB alloys reported by ter and $CuK\alpha$ radiation. Magnetic measurements were Clemente et al. [4] for a near stoichiometric melt spun carried out at room temperature using a Manics DSM. $Nd₂Fe₁₄B$ and by Coehoorn et al. [5] for a melt spun NdFeB material which mixed the hard magnetic $Nd₂Fe₁₄B$ with soft magnetic Fe₃B and α -Fe. Subsequent studies by **3. Results and discussion** Ding et al. [6] using $Sm_2Fe_{17}N_x$ with an excess of α -Fe showed that it was possible to obtain exchange-coupled All samples were investigated after mechanical alloying materials showing enhanced remanences by mechanical in order to determine that the materials had become alloying. Materials with BH_{max} values of 20 MGOe were amorphous. X-ray diffraction patterns indicated only very produced, similar energy products have been obtained by broad peaks of low intensity similar to those obs produced, similar energy products have been obtained by

1. Introduction 1. Introduction Although the literature contains some reports of mechanical alloying of SmCo-based alloys in order to develop Hard magnetic nanocrystalline materials consisting of a isotropic [9] and exchange-coupled [10,11] magnets this

where $x=0.95-1.25$, were mechanically alloyed for 3000

melt spining $Nd_2Fe_{14}B+\alpha$ -Fe ribbons [7,8]. other workers [10]. The material exhibited a dark grey colour and the particle size and morphology were investi- *Corresponding author. gated using a JEOL 5800 scanning electron microscope

(SEM). A relatively narrow particle size distribution with most particles falling within the range $10-30 \mu m$ was observed, there is little evidence for significant amounts of debris. A close up view of one particle (see Fig. 1) confirms that each particle is basically spherical in shape and appears to be made up of layers of material in agreement with previous observations for mechanically alloyed materials [12].

During the process of mechanical alloying a significant amount of alloying occurs not just between the particles themselves but also between the material and the mechanical alloying chamber. As a result of this it is often not possible to retrieve all of the initial charge. In the course of these experiments it was found that 2 or 3 g of material from the initial charge of 15 g could not be removed using any non abrasive technique. In order to make certain that
no significant change in the chemical composition was
taking place as a result of preferential alloying of one element to the walls of the chamber, chemical analysis was carried out on all of the mechanically alloyed samples. The The slight increase in the degree of pick up with higher graph, Fig. 2 shows slight preferential loss of samarium in rare earth containing materials is consistent with the the order of 1 wt. %. formation of Sm based oxides. Measurements relating to

powders of high reactivity, oxygen analysis was carried not undertaken, experience from earlier studies on similar out on the samples mechanically alloyed and heat treated materials has shown that significant quantities are not at 875°C. The graph, Fig. 3 shows oxygen levels in the observed. region of 0.4–0.6 wt. %, typical of that observed in other The mechanically alloyed samples in the range $Sm_cCo₅$ forms of processing rare earth-transition metal materials. where $x=0.95-1.25$ were each heat treated for 2 h at the

Fig. 1. Single particle of mechanically alloyed $Sm_{1,15}Co_5$. and heat treated at 875°C for 2 hours.

Since the experimental work involved dealing with fine the amount of iron picked up during the experiment were

following temperatures, 575° C, 675° C, 775° C and 875° C, the heating and cooling rates were 5° C/min and the furnace tube was filled with argon at a pressure of 1 bar after pre-evacuation.

X-ray diffraction was carried out using $CuK\alpha$ radiation. In Fig. 4 the traces from the as-mechanically alloyed sample of $Sm_{1,15}Co_5$ as well as $Sm_{1,15}Co_5$ samples heat treated at 575° C, 675° C, 775° C and 875° C are shown. The un-heat treated sample shows a very flat trace with broad

Fig. 3. Oxygen content of samples mechanically alloyed for 3000 mins

phase ($P6/mmm$) with all the strongest peaks from $d=$ 0.2929 nm to 0.1556 nm being accounted for. A significant Further studies into the material's structure and grain amount of cobalt is retained during the development of the size will be undertaken in subsequent studies. $\rm SmCo_5$ phase at high temperatures, it is possible that this Ding et al. [10] carried out an investigation with SmCo could result from oxidation during heat treatment. There is powders somewhat less rich in Sm than those used in this little evidence for the development of much 2:17 phase. series of experiments. In addition the enhanced remanence

powder was found to be magnetically soft, and the coupling between 2-17 and 1-5 phases, which is different saturation magnetisation was found to decrease with to that observed in this study. Smith et al. [11] in their increasing amounts of samarium. The effect of the heat study of SmCoFe alloys again observed a different structreatments on the room temperature magnetic properties is ture with the hard magnetic phase being a 1-7 or 2-7 phase shown in Fig. 5. The graph shows the variation of and the soft phase being bcc FeCo. The differences in coercivity, H_c , as a result of the different heat treatments. compositions make it hard to draw direct comparisons The low samarium content samples exhibit very poor between these studies and the present one, however it

Fig. 5. The effect of heat treatment temperature on the coercivity of samples of Sm, Co₅. Fig. 6. The effect of heat treatment temperature on the ratio M_r/M_s .

coercivities of less than 1 kOe. Increasing the amount of samarium results in the development of substantial coercivity in the order of 8 kOe. Further increase in the amount of added Sm results in further increase in the materials resistance to demagnetisation. The largest coercivity, 9.8 kOe, was obtained with the $Sm_{1.25}Co_5$ alloy heat treated at $600-700^{\circ}$ C. Coercivity was observed to be strongly linked to the Sm content of the powder.

In order to assess the degree of exchange-coupling taking place in these materials a graph of M_r/M_s vs heat treatment temperature was plotted, Fig. 6. The value of M_s being obtained from the first quadrant of the hysteresis loop. All the materials were observed to saturate and the Fig. 4. X-ray diffraction traces from samples of mechanically alloyed curves exhibited a smooth hysteresis loop. The graph again $\rm Sm_{1.15}Co_5$ heat treated at 575°C, 675°C, 775°C and 875°C. Shows very poor results for the low Sm materials $Sm_{0.95}Co_5$ and $Sm_{1.00}Co_5$ which have M_r/M_s values of only \sim 0.1. An increase in the Sm content to Sm_{1.05}Co₅ low intensity peaks characteristic of a mixture of sees a very significant jump in the maximum value of nanocrystalline Co and a possible amorphous phase, the M_r/M_s to ~0.69 for the material heat treated at 775°C. peaks at 20=475° and 44.5° can be indexed to Co (*P63*/ Similar although slightly lower values were obtained f Similar although slightly lower values were obtained for *mmc*). Low temperature heat treatment at 575° C and the $Sm_{1.15}Co_5$ alloy. Increasing the samarium content 6758C, appears to develop an intermediate SmCo structure further, leads to a decrease in the maximum value for which has not been identified but at higher temperatures M_r / M_s of ~ 0.5 , i.e. half the value of the saturation 775°C and 875°C there is good evidence for the SmCo₅ polarisation and in accordance with that predicted from phase (*P6/mmm*) with all the strongest peaks from $d =$ simple theory for an isotropic material [13].

Prior to any heat treatment the as mechanically alloyed they report appears to come about as a result of exchange between these studies and the present one, however it does

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Coercivity appears to develop as a consequence of the formation of the $SmCo₅$ phase and the remanence enhancement as a result of exchange-coupling between this $SmCo₅$ phase and cobalt.

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